

## **Purpose**

The goal of this activity is to learn how to make casein plastic from milk and explore the properties of plastic.

**Skills:** This activity will help you practice these skills:

- Describe the differences between starting and final ingredients
- Observation skills

**Knowledge:** This activity will also help you to become familiar with the knowledge that is needed in fields like chemistry or chemical engineering:

- Plastic polymers
- Protein bonding and denaturing

### **Task**

- Today you will:
  - Listen to an introduction about bonds between molecules
  - Mix ingredients together to form a polymer
  - Make observations about the reaction that forms the polymer.
  - Brainstorm why different ingredients might lead to different observations.
  - Share your ideas!
- As you are completing the experiment consider these questions:
  - Have you ever seen milk separate into thick chunks and liquid? Why did that happen?
  - Why does the milk form curd when the acid is added? What do you think the curds are made of?
  - o What aspects of the acid cause the changes in the milk?
  - Why does the kneaded polymer look and feel different from the unkneaded polymer?



- Discussion questions to consider after the experiment:
  - What would happen if room temperature or cold milk is used instead of heated milk?
  - How would increasing or decreasing the amount of acid affect how much polymer you get?
  - What are other household acids that you could use for this experiment?

## **Tips for Success:**

- Focus on vivid descriptions of the reaction and the final product.
  - o Use descriptive words such as "cloudy," "opaque," "shiny," etc.
  - Compare what you see to common things such as the thickness of syrup, oil, or melted butter.
  - Feel free to draw pictures to capture what is happening.
  - Focus your descriptions in a way that someone who has not conducted the experiment can visualize what you saw.
- Some good descriptions you can use:
  - o Texture: rough, smooth, bumpy, sticky, tacky
  - o Color: white, gray, clear, shadowy, uneven
  - o Odor: sour, sharp, floral, yeasty
  - o Size: "grain of sand", "size of a dime"
  - o Shape: uneven, round, oblong, stringy
  - Example: "the curds formed after the reaction were small about the size of rice, not see through, and white. The liquid looked like chicken broth, but it smelled sour like vinegar. The polymer was bumpy and sticky at first but after kneading it wasn't sticky, but it was still bumpy."
- Excellent work includes writing down your observations at all points of the reaction:
  - What everything looks like before you mix it
  - What happens immediately when you add them together
  - What (and how long) the reaction happens
  - What it looks like when it is done
  - What changes happen as you knead the polymer



# Instructions

1.	Read all the directions first.				
	1.1.	Read the directions through to the end.			
	1.2.	Get the observation table out and ready.			
2.	Obtain	materials.			
	2.1.	Get 1 cup of hot milk and 4 teaspoons of your assigned acid. Everyone will get a different acid, so be sure to take good notes so you can share your observations later!			
	2.2.	Prepare cheesecloth by folding 4 layers on top of itself or fold about 4-5 paper towels in a stack for later use.			
	2.3.	Write down your observations about your starting ingredients.			
3.	Combin	ne liquids to create a reaction.			
	3.1.	Pour the 4 teaspoons of acid into the milk and begin mixing with a spoon.			
	3.2.	Observe the changes happening in the liquid and record them on your worksheet. Pay attention to the color, texture, odor, size, and shape of the milk/curds as well as the speed of the changes.			
	3.3.	Mix the solution for about 30 seconds until you see no more curds forming.			
	3.4.	Let the mixture sit for about 1 minute to let the curds settle to the bottom (this makes it easier to remove them).			
	3.5.	What do the curds look like? How many are there? Record your observations in your worksheet.			
4.	Remov	e and dry curds.			
	4.1.	Use the cheesecloth or gauze to filter the curds from the liquid. Find a partner to help by holding the cloth above an empty cup. Take your cup and pour the liquid into the cup through the cloth. Finally take all sides of the cloth and squeeze the extra liquid out into the cup.			
	4.2.	Place the curds in a pile on a paper towel and pat dry if needed.			
	4.3.	What do the dry curds look like? What do they feel like?			
5.	Knead	polymer.			
	5.1.	Begin to massage and squeeze the polymer slowly. This is called "kneading". Some of the polymer will stick to your hands but as you keep kneading it, the polymer will come off.			
	5.2.	Keep kneading for 2-4 minutes or until it is a smooth, solid mass.			
	5.3.	How did the curds feel? Did that change as you kneaded them? How did their appearance change? Record your observations in your table.			



O.	Clean up					
	6.1.	Set polymer aside to dry for 24-48 hours				
	6.2.	Throw cup with leftover milk, cheesecloth/gauze, and paper towels in the trash (NOT down the drain)				
	6.3.	Wipe down area with a clean cloth and wash your hands.				
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7.	<b>Think-F</b>	Pair-Share  Brainstorm different ways to do this experiment. Think about how you could change the milk, the				
7.	7.1.	Pair-Share  Brainstorm different ways to do this experiment. Think about how you could change the milk, the acid, or the amount of plastic you get. Hold on to these ideas for the class discussion.  Find a partner who used a different acid and compare observations of your polymers. Write down similarities and differences of your plastics on your worksheet				



#### Worksheet

The goal of this activity is to learn how to make casein plastic from milk and explore the properties of plastic. This worksheet is where you will record all your observations.

Description of starting ingredients:

Texture	Color	Odor	Size	Shape	Other Observations
Milk	Milk	Milk	Milk	Milk	Milk
Acid	Acid	Acid	Acid	Acid	Acid
, told	71010	, told	71014	71010	7.00

Description of reaction: Describe what you see as you add the acid into the milk; focus on texture, color, odor, size, and shape descriptions. Note the speed of the reaction.

Color	Odor	Size	Shape	Speed
	Color	Color Odor	Color Odor Size	Color Odor Size Shape



## Description of final product:

Texture	Color	Odor	Size	Shape	Other Observations
Polymer	Polymer	Polymer	Polymer	Polymer	Polymer
Liquid	Liquid	Liquid	Liquid	Liquid	Liquid

**Think-Pair-Share:** Describe similarities and differences between your polymer and your partner's polymer.

Differences

Final explanations and observations: Any comments or explanations of the experiment.